

Cambridge International Examinations

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/32

Paper 3 Core Theory May/June 2017

MARK SCHEME
Maximum Mark: 80

Published

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[Turn over

Question	Answer	Marks
1(a)	lines drawn from Enzymes to are biological catalysts; are usually not active at low temperatures; are protein molecules;	3
1(b)	large / insoluble / food molecules are broken down; into small / soluble molecules / so they can be absorbed;	2
1(c)	glycogen; starch;	2
1(d)(i)	Benedict's (test); red colour produced;	2
1(d)(ii)	no reaction ; because enzymes become inactive at high temperatures ;	2
1(e)	chlorophyll; light;	2

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2(c)(ii)

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Question	Answer	Marks
2(a)(i)	potassium / K lithium / Li sodium / Na ;;	2
2(a)(ii)	hydrogen / H ₂ ;	1
2(a)(iii)	turns blue and stays blue / no change ;	1
2(b)(i)	magnesium / Mg ;	1
2(b)(ii)	copper / Cu ;	1
2(b)(iii)	(too) dangerous / (risk of) explosion ;	1
2(c)(i)	resists corrosion / does not rust ;	1

stronger / does not get damaged;

Question		Answer		Marks
3(a)(i)				2
	name of force	letter on Fig. 1.1		
	driving force	Α		
	frictional force	С		
	lifting force	В		
	weight	D		
	one mark for each two correct ;;			
3(a)(ii)	(Force B is 500 000 N) no mark constant height; forces (B and D) are balanced;			1
3(a)(iii)	1. A / driving force ; 2. B / lifting force ;			2
3(b)(i)	600 km/h = 600 000 / 3600 m/s = 167 m/s;			1
3(b)(ii)	time (= distance / speed) = 2700 / 600 = 4.5 h			1
3(c)	loss of kinetic energy ; loss of (gravitational) potential energy ;			2
3(d)	any variation on this shape that goes from the origin to a r horizontal section at constant maximum speed;	naximum and returns to sp	eed = 0 ;	2

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Question	Answer	Marks
4(a)(i)	A closes and B opens ;	1
4(a)(ii)	to prevent backflow of blood;	1
4(b)(i)	any suitable flight or fight situation described ;	1
4(b)(ii)	destroyed by the liver ;	1
4(c)	transport of oxygen / haemoglobin; transport of blood cells / ions / soluble nutrients / named soluble nutrient / hormones / carbon dioxide;	2

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Question	Answer	Marks
5(a)(i)	Fractional distillation ;	1
5(a)(ii)	no new substance made / involves only changes of state ;	1
5(a)(iii)	cooking / heating allow bottling / bottled gas ;	1
5(b)(i)	methane;	1
5(b)(ii)	(atoms) five / 5 and (elements) two / 2;	1
5(b)(iii)	C atom joined to 4 H atoms by single bonds ; allow correct dot-and-cross diagrams	1
5(c)	coal;	1

Question				Answer	Marks
6(a)(i)	conduction;			1	
6(a)(ii)	insulation (in oute	r layer of aircraft)/	make aircraft o	out of bad (thermal) conductor / owtte ;	1
6(b)(i)	(Z – no mark) gas molecules far	apart / not touchin	g;		1
6(b)(ii)	ice / (frozen) water ; water from fuel combustion freezing / condensing in very cold air ;			2	
6(c)	gamma radiation	visible light	micro- waves ;	radio waves ;	2
6(d)	(pitch) low ; (amplitude) (very)	high ;			2

Question	Answer	Marks
7(a)(i)	for <u>respiration</u> ;	1
7(a)(ii)	diffusion;	1
7(a)(iii)	from the (water) plants ;	1
7(b)(i)	food web completed as shown ;	2
	small animals — fish	
	algae water plants	
	arrows in the correct direction ;	
7(b)(ii)	small animals ; water plants / algae ;	2

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Question	Answer	
8(a)(i)	filtration;	1
8(a)(ii)	kill microbes / sterilise (water);	1
8(a)(iii)	(damp)-litmus (paper) ; turns white / bleached ;	2
8(b)(i)	chlorine + hydrogen → hydrogen chloride ; LHS either order ;	1
8(b)(ii)	covalent ; share (pair of) electrons ;	2
8(b)(iii)	HC1;	1
8(c)(i)	anode;	1
8(c)(ii)	copper;	1
8(c)(iii)	copper chloride solution / aqueous copper chloride ;	1

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Question	Answer	Marks
9(a)	correct symbols for ammeter and lamp; correct symbol for variable resistor; all shown components connected in series, any order;	3
9(b)	resistance = V/I ; (total resistance) = $2.4/0.6$ (= 4Ω) ; resistance of one lamp = $2(\Omega)$;	3
9(c)	(increase – no mark) (total resistance less) so current increases ;	1